

# APPENDIX C

## Useful Information

This appendix is broken into several tables.

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Symbol	Meaning	Best Value	Approximate Value
$c$	Speed of light in vacuum	$2.99792458 \times 10^8$ m/s	$3.00 \times 10^8$ m/s
$G$	Gravitational constant	$6.67408(31) \times 10^{-11}$ N · m <sup>2</sup> /kg <sup>2</sup>	$6.67 \times 10^{-11}$ N · m <sup>2</sup> /kg <sup>2</sup>
$N_A$	Avogadro's number	$6.02214076 \times 10^{23}$	$6.02 \times 10^{23}$
$k$	Boltzmann's constant	$1.380649 \times 10^{-23}$ J/K	$1.38 \times 10^{-23}$ J/K
$R$	Gas constant	$8.3144621(75)$ J/mol · K	$8.31$ J/mol · K = $1.99$ cal/mol · K $= 0.0821$ atm · L/mol · K
$\sigma$	Stefan-Boltzmann constant	$5.670373(21) \times 10^{-8}$ W/m <sup>2</sup> · K	$5.67 \times 10^{-8}$ W/m <sup>2</sup> · K
$k$	Coulomb force constant	$8.987551788... \times 10^9$ N · m <sup>2</sup> /C <sup>2</sup>	$8.99 \times 10^9$ N · m <sup>2</sup> /C <sup>2</sup>
$q_e$	Charge on electron	$-1.602176634 \times 10^{-19}$ C	$-1.60 \times 10^{-19}$ C
$\epsilon_0$	Permittivity of free space	$8.854187817... \times 10^{-12}$ C <sup>2</sup> /N · m <sup>2</sup>	$8.85 \times 10^{-12}$ C <sup>2</sup> /N · m <sup>2</sup>
$\mu_0$	Permeability of free space	$4\pi \times 10^{-7}$ T · m/A	$1.26 \times 10^{-6}$ T · m/A
$h$	Planck's constant	$6.62607015 \times 10^{-34}$ J · s	$6.63 \times 10^{-34}$ J · s

TABLE C1 Important Constants <sup>1</sup>

<sup>1</sup> Stated values are according to the National Institute of Standards and Technology Reference on Constants, Units, and Uncertainty, [www.physics.nist.gov/cuu](http://www.physics.nist.gov/cuu) (<http://www.physics.nist.gov/cuu>) (accessed May 18, 2012). Values in parentheses are the uncertainties in the last digits. Numbers without uncertainties are exact as defined.

Symbol	Meaning	Best Value	Approximate Value
$m_e$	Electron mass	$9.10938291(40) \times 10^{-31} \text{ kg}$	$9.11 \times 10^{-31} \text{ kg}$
$m_p$	Proton mass	$1.672621777(74) \times 10^{-27} \text{ kg}$	$1.6726 \times 10^{-27} \text{ kg}$
$m_n$	Neutron mass	$1.674927351(74) \times 10^{-27} \text{ kg}$	$1.6749 \times 10^{-27} \text{ kg}$
u	Atomic mass unit	$1.660538921(73) \times 10^{-27} \text{ kg}$	$1.6605 \times 10^{-27} \text{ kg}$

**TABLE C2** Submicroscopic Masses <sup>2</sup>

<b>Sun</b>	mass	$1.99 \times 10^{30} \text{ kg}$
	average radius	$6.96 \times 10^8 \text{ m}$
	Earth-sun distance (average)	$1.496 \times 10^{11} \text{ m}$
<b>Earth</b>	mass	$5.9736 \times 10^{24} \text{ kg}$
	average radius	$6.376 \times 10^6 \text{ m}$
	orbital period	$3.16 \times 10^7 \text{ s}$
<b>Moon</b>	mass	$7.35 \times 10^{22} \text{ kg}$
	average radius	$1.74 \times 10^6 \text{ m}$
	orbital period (average)	$2.36 \times 10^6 \text{ s}$
	Earth-moon distance (average)	$3.84 \times 10^8 \text{ m}$

**TABLE C3** Solar System Data

Prefix	Symbol	Value	Prefix	Symbol	Value
tera	T	$10^{12}$	deci	d	$10^{-1}$
giga	G	$10^9$	centi	c	$10^{-2}$
mega	M	$10^6$	milli	m	$10^{-3}$
kilo	k	$10^3$	micro	$\mu$	$10^{-6}$
hecto	h	$10^2$	nano	n	$10^{-9}$

**TABLE C4** Metric Prefixes for Powers of Ten and Their Symbols

<sup>2</sup> Stated values are according to the National Institute of Standards and Technology Reference on Constants, Units, and Uncertainty, [www.physics.nist.gov/cuu](http://www.physics.nist.gov/cuu) (<http://www.physics.nist.gov/cuu>) (accessed May 18, 2012). Values in parentheses are the uncertainties in the last digits. Numbers without uncertainties are exact as defined.

Prefix	Symbol	Value	Prefix	Symbol	Value
deka	da	$10^1$	pico	p	$10^{-12}$
—	—	$10^0 (= 1)$	femto	f	$10^{-15}$

**TABLE C4** Metric Prefixes for Powers of Ten and Their Symbols

Alpha	A	$\alpha$	Eta	H	$\eta$	Nu	N	$\nu$	Tau	T	$\tau$
Beta	B	$\beta$	Theta	$\Theta$	$\theta$	Xi	$\Xi$	$\xi$	Upsilon	$\Upsilon$	$\upsilon$
Gamma	$\Gamma$	$\gamma$	Iota	I	$\iota$	Omicron	O	$\circ$	Phi	$\Phi$	$\phi$
Delta	$\Delta$	$\delta$	Kappa	K	$\kappa$	Pi	$\Pi$	$\pi$	Chi	X	$\chi$
Epsilon	E	$\epsilon$	Lambda	$\Lambda$	$\lambda$	Rho	P	$\rho$	Psi	$\Psi$	$\psi$
Zeta	Z	$\zeta$	Mu	M	$\mu$	Sigma	$\Sigma$	$\sigma$	Omega	$\Omega$	$\omega$

**TABLE C5** The Greek Alphabet

	Entity	Abbreviation	Name
<b>Fundamental units</b>	Length	m	meter
	Mass	kg	kilogram
	Time	s	second
	Current	A	ampere
<b>Supplementary unit</b>	Angle	rad	radian
<b>Derived units</b>	Force	$N = kg \cdot m/s^2$	newton
	Energy	$J = kg \cdot m^2/s^2$	joule
	Power	$W = J/s$	watt
	Pressure	$Pa = N/m^2$	pascal
	Frequency	$Hz = 1/s$	hertz
	Electronic potential	$V = J/C$	volt
	Capacitance	$F = C/V$	farad
	Charge	$C = s \cdot A$	coulomb

**TABLE C6** SI Units

	Entity	Abbreviation	Name
	Resistance	$\Omega = V/A$	ohm
	Magnetic field	$T = N/(A \cdot m)$	tesla
	Nuclear decay rate	$Bq = 1/s$	becquerel

**TABLE C6** SI Units

Length	1 inch (in.) = 2.54 cm (exactly)
	1 foot (ft) = 0.3048 m
	1 mile (mi) = 1.609 km
Force	1 pound (lb) = 4.448 N
Energy	1 British thermal unit (Btu) = $1.055 \times 10^3$ J
Power	1 horsepower (hp) = 746 W
Pressure	1 lb/in <sup>2</sup> = $6.895 \times 10^3$ Pa

**TABLE C7** Selected British Units

Length	1 light year (ly) = $9.46 \times 10^{15}$ m
	1 astronomical unit (au) = $1.50 \times 10^{11}$ m
	1 nautical mile = 1.852 km
	1 angstrom ( $\text{\AA}$ ) = $10^{-10}$ m
Area	1 acre (ac) = $4.05 \times 10^3$ m <sup>2</sup>
	1 square foot (ft <sup>2</sup> ) = $9.29 \times 10^{-2}$ m <sup>2</sup>
	1 barn (b) = $10^{-28}$ m <sup>2</sup>
Volume	1 liter (L) = $10^{-3}$ m <sup>3</sup>
	1 U.S. gallon (gal) = $3.785 \times 10^{-3}$ m <sup>3</sup>
Mass	1 solar mass = $1.99 \times 10^{30}$ kg
	1 metric ton = $10^3$ kg
	1 atomic mass unit (u) = $1.6605 \times 10^{-27}$ kg

**TABLE C8** Other Units

Time	$1 \text{ year (y)} = 3.16 \times 10^7 \text{ s}$
	$1 \text{ day (d)} = 86,400 \text{ s}$
Speed	$1 \text{ mile per hour (mph)} = 1.609 \text{ km/h}$
	$1 \text{ nautical mile per hour (naut)} = 1.852 \text{ km/h}$
Angle	$1 \text{ degree } (\circ) = 1.745 \times 10^{-2} \text{ rad}$
	$1 \text{ minute of arc } ('') = 1/60 \text{ degree}$
	$1 \text{ second of arc } ('') = 1/60 \text{ minute of arc}$
Energy	$1 \text{ grad} = 1.571 \times 10^{-2} \text{ rad}$
	$1 \text{ kiloton TNT (kT)} = 4.2 \times 10^{12} \text{ J}$
	$1 \text{ kilowatt hour (kW} \cdot h) = 3.60 \times 10^6 \text{ J}$
	$1 \text{ food calorie (kcal)} = 4186 \text{ J}$
	$1 \text{ calorie (cal)} = 4.186 \text{ J}$
Pressure	$1 \text{ electron volt (eV)} = 1.60 \times 10^{-19} \text{ J}$
	$1 \text{ atmosphere (atm)} = 1.013 \times 10^5 \text{ Pa}$
	$1 \text{ millimeter of mercury (mm Hg)} = 133.3 \text{ Pa}$
Nuclear decay rate	$1 \text{ curie (Ci)} = 3.70 \times 10^{10} \text{ Bq}$

**TABLE C8** Other Units

Circumference of a circle with radius $r$ or diameter $d$	$C = 2\pi r = \pi d$
Area of a circle with radius $r$ or diameter $d$	$A = \pi r^2 = \pi d^2/4$
Area of a sphere with radius $r$	$A = 4\pi r^2$
Volume of a sphere with radius $r$	$V = (4/3)(\pi r^3)$

**TABLE C9** Useful Formulae

